## **Example Purpose and Conclusion Chemistry**

PURPOSE: The purpose of this experiment is to determine the mass ratio that magnesium and oxygen react in (What) by heating the magnesium in a crucible and making careful mass measurements (HOW).

CONCLUSION: The mass ratio that magnesium and oxygen react in was determined (WHAT) by heating the magnesium in a crucible and making careful mass measurements. (HOW) The ratio was calculated to be 1.72gMg/1.00gO (RESULT). **This represents an error of** 13.20% **compared to the actual** ratio of 1.52. **One source of error may be** that the magnesium was not heated long enough so that the reaction was complete resulting in a ratio with too much magnesium (SOURCE).

## A few things to notice:

- 1. No pronouns were used
- 2. WHAT, HOW are all in one sentence
- 3. HOW: is NOT a repeated summary of the whole procedure it is one sentence describing the METHOD used.
- 4. The RESULTS answers the WHAT. This is not DATA! This will not be found in your data table yet in the calculations & analysis you make OF the data.